# organic compounds

Acta Crystallographica Section C Crystal Structure Communications ISSN 0108-2701

# Methyl (±)-1-ethyl-2-hydroxy-4-(4-methoxybenzoyl)-5-(4-methoxyphenyl)-3-oxo-2,3-dihydro-1*H*pyrrole-2-acetate

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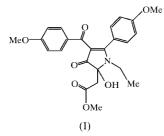
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Received 20 May 2003 Accepted 20 August 2003 Online 23 September 2003

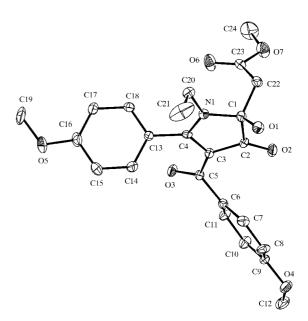
The title compound,  $C_{24}H_{25}NO_7$ , is a racemic mixture of 2,3dihydro-1*H*-pyrrol-3-ones. It crystallizes in the triclinic system, space group *P*1, with *Z* = 2. The asymmetric unit contains two enantiomorphic molecules and the structure is stabilized by hydrogen-bond contacts.

# Comment

Pyrrolones have received considerable attention as a result of the presence of the lactam ring in some antibiotics, in bile pigments and in the natural alkaloid jatropham, which shows inhibitory activity towards P-338 lymphocytic leukaemia (Jimenez *et al.*, 1988; Gossauer, 1983; Wiedhopf *et al.*, 1973). Among these chemical structures, 2,3-dihydro-1*H*-pyrrol-3one derivatives (Weigele *et al.*, 1972, 1976; Gelin & Gelin, 1979; Chantegrel & Gelin, 1978) exhibit remarkable antimicrobiological and pharmacological activities (Koz'minykh *et al.*, 1992, 1996, 2002). In the light of this, we have synthesized the title compound, (I), which consists of two enantiomers, and present its X-ray crystal structure.



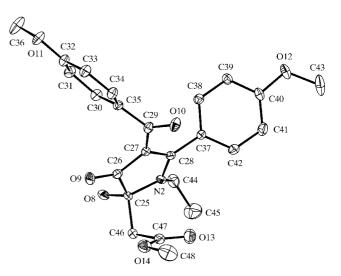
The two molecules in the asymmetric unit of (I), (Ia) and (Ib), are enantiomorphs of each other, as shown in Figs. 1 and 2 and *via* the torsion angles in Table 1. The bond lengths are



## Figure 1

The molecular structure of molecule (I*a*), showing the atom-numbering scheme and with displacement ellipsoids drawn at the 20% probability level. H atoms have been omitted for clarity.

almost identical in both molecules and are comparable with those in related structures (Allen *et al.*, 1987). In molecule (I*a*), the angles between planes *A* (N1/C1–C4), *B* (C6–C11) and *C* (C13–C18) are A/B = 57.4 (2), A/C = 60.0 (2) and B/C = 85.5 (2)°. In molecule (I*b*), the equivalent planes are *D* (N2/C25–C28), *E* (C30–C35) and *F* (C37–C42), and the angles between them are D/E = 69.4 (2), D/F = 54.7 (2) and E/F = 89.0 (2)°. These values are similar to those in 2(*R*)-hydroxy-4-(4-methoxybenzoyl)-2-methoxycarbonylmethyl-5-(4-methoxyphenyl)-1-(2-methylphenyl)-2,3-dihydro-1*H*-3-pyrrolone (Öztürk *et al.*, 2003). The angles between the planes of



### Figure 2

The molecular structure of molecule (Ib), showing the atom-numbering scheme and displacement ellipsoids drawn at the 20% probability level. H atoms have been omitted for clarity.

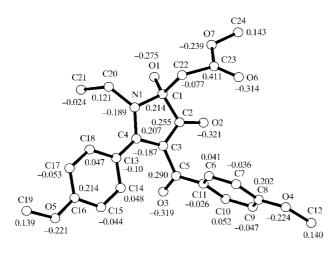


Figure 3 The CNDO-calculated charge distribution for (I).

similar rings of the two enantiomorphic molecules of (I) are not identical, with values of A/D = 41.6 (2), B/E = 68.2 (2) and  $C/F = 51.1 \ (2)^{\circ}.$ 

We have carried out CNDO (complete neglect of differential overlap) quantum mechanical calculations on (I), and the calculated charge distribution on the molecule of (I) in the gas phase is shown in Fig. 3. The dipole moment of (I) is ~17.51  $\times 10^{-30}$  Cm. The highest occupied and lowest unoccupied molecular orbital energy levels are -10.1370 and 1.1187 eV, respectively.

The crystal structure of (I) reveals a number of threedimensional  $O-H \cdots O$  and  $C-H \cdots O$  short hydrogen-bond contacts, details of which are given in Table 2.

# **Experimental**

A mixture of 4-(4-methoxybenzoyl)-2-[(Z)-methoxycarbonylmethylene]-5-(4-methoxyphenyl)-2,3-dihydro-3-furanone (2.213 g, 5 mmol) and ethylamine (70% in water; 0.322 g, 0.40 ml, 5 mmol) in benzene (25 ml) was refluxed for 30 min. The solid remaining after evaporation of the solvent was recrystallized from ethanol to give white crystals of (I) (yield: 1.54 g, 70%; m.p. 435 K). Solvents were dried by refluxing with the appropriate drying agent and were distilled before use. Melting points were determined on an Electrothermal 9200 apparatus and are uncorrected. Elemental analysis was performed with a Carlo-Erba 1108 Elemental Analyzer. The FT-IR spectrum was measured on a Jasco Plus Model 460 spectrometer. The <sup>1</sup>H and <sup>13</sup>C NMR spectra were obtained on a Gemini-Varian 200 MHz instrument. Spectroscopic analysis, FT–IR (KBr pellet,  $\nu$ , cm<sup>-1</sup>): 3212.34 (broad OH), 1733.69 (C12=O), 1678.69 (C8=O), 1674.39 (C4=O), 1251.58 (C12-O-C13); <sup>1</sup>H NMR (CDCl<sub>3</sub>, p.p.m.): 1.095-1.167 (s, 3H, CH<sub>3</sub>), 2.940–3.211 (q, 2H, CH<sub>2</sub>C=O), 3.300–3.604 (q, 2H, NCH<sub>2</sub>), 3.884, 3.780, 3.797 (s, 9H, OCH<sub>3</sub>), 5.942 (s, 1H, OH), 6.731-7.700 (m, 8H, Ar-H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, p.p.m.): 17.474 (C7), 40.230 (C11), 42.588 (C6), 53.904, 57.216 (C13, C10, C9), 90.746 (C5), 112.392–164.865 (C=C, aromatic and aliphatic), 182.945 (C12=O), 188.924 (C8=O), 196.037 (C4=O). Analysis calculated for C<sub>24</sub>H<sub>25</sub>-NO<sub>7</sub>: C 65.59, N 3.19, H 5.73%; found: C 65.41, N 3.30, H 5.72%.

| C <sub>24</sub> H <sub>25</sub> NO <sub>7</sub> | Z = 2  |
|---|--|
| $M_r = 439.45$                                  | $D_x = 1.292 \text{ Mg m}^{-3}$              |
| Triclinic, P1                                   | Mo $K\alpha$ radiation                       |
| a = 10.160 (5)  Å                               | Cell parameters from 2644                    |
| b = 10.785(5) Å                                 | reflections                                  |
| c = 11.942(5)  Å                                | $\theta = 4.6-49.1^{\circ}$                  |
| $\alpha = 66.369 \ (5)^{\circ}$                 | $\mu = 0.10 \text{ mm}^{-1}$                 |
| $\beta = 75.928 \ (5)^{\circ}$                  | T = 293  K                                   |
| $\gamma = 72.005 \ (5)^{\circ}$                 | Block, colourless                            |
| $V = 1129.4 (9) \text{ Å}^3$                    | $0.52 \times 0.26 \times 0.20 \ \mathrm{mm}$ |

#### Data collection

Siemens SMART CCD areadetector diffractometer  $\omega$  scans Absorption correction: refined on  $\Delta F$  [Parkin et al., 1995; cubic fit to  $\sin(\theta)/\lambda$  with 24 parameters]  $T_{\rm min}=0.952,\ T_{\rm max}=0.981$ 

### Refinement

Refinement on  $F^2$ R(F) = 0.066 $wR(F^2) = 0.127$ S = 1.116035 reflections 577 parameters H-atom parameters constrained

| Block, colourless                         |
|---|
| $0.52 \times 0.26 \times 0.20 \text{ mm}$ |
|   |
|   |
|   |
| 6035 measured reflections                 |
| 6035 independent reflection               |
| 4369 reflections with $I > 2\sigma$       |

(I) $\theta_{\rm max} = 28.3^{\circ}$  $h = -13 \rightarrow 13$  $k = -14 \rightarrow 11$  $l = -15 \rightarrow 15$ 

| $w = 1/[\sigma^2(F_o^2) + (0.0372P)^2]$                    |
|--|
| + 0.2621P]   |
| where $P = (F_o^2 + 2F_c^2)/3$                             |
| $(\Delta/\sigma)_{\rm max} = 0.002$                        |
| $\Delta \rho_{\rm max} = 0.22 \text{ e } \text{\AA}^{-3}$  |
| $\Delta \rho_{\rm min} = -0.19 \ {\rm e} \ {\rm \AA}^{-3}$ |
|  |

#### Table 1 Selected torsion angles (°).

| C4-N1-C1-O1 | 121.1 (4)  | C28-N2-C25-O8 | 127.2 (4) |
|-------------|------------|---------------|-----------|
| C4=N1=C1=01 | -121.1 (4) | C28-N2-C25-08 | 127.2 (4) |

#### Table 2

Hydrogen-bonding and short intra/intermolecular contact geometry (A, °).

| $D - H \cdots A$             | $D-\mathrm{H}$ | $H \cdot \cdot \cdot A$ | $D \cdots A$ | $D - \mathbf{H} \cdots A$ |
|------------------------------|----------------|-------------------------|--------------|---------------------------|
| $O1 - H1 \cdots O10^i$       | 0.82           | 2.04                    | 2.844 (5)    | 168                       |
| $O8-H8A\cdots O3$            | 0.82           | 1.95                    | 2.754 (5)    | 169                       |
| $C12-H12C\cdots O1^{ii}$     | 0.96           | 2.54                    | 3.434 (7)    | 155                       |
| C19−H19A···O9 <sup>iii</sup> | 0.96           | 2.48                    | 3.211 (7)    | 133                       |
| $C24 - H24A \cdots O4^{iv}$  | 0.96           | 2.54                    | 3.440 (10)   | 156                       |
| C44-H44AO8                   | 0.97           | 2.45                    | 2.869 (6)    | 106                       |
| $C45-H45B\cdots O11^{v}$     | 0.96           | 2.56                    | 3.471 (10)   | 159                       |

Symmetry codes: (i) 1 + x, y, z; (ii) x, y - 1, z; (iii) x, y, 1 + z; (iv) x - 1, 1 + y, z; (v) x, y - 1, 1 + z.

H atoms were introduced at calculated positions and refined as riding atoms, with bond lengths of 0.82 (O-H), 0.93 (C-H aromatic), 0.96 (CH<sub>3</sub>) and 0.97 Å (CH<sub>2</sub>), and with  $U_{iso}(H) =$  $1.2U_{eq}$ (parent atom) or  $1.5U_{eq}$ (methyl C and hydroxyl O).

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SIR97 (Altomare et al., 1999); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP-3 for Windows (Farrugia, 1997); software used to prepare material for publication: PARST (Nardelli, 1995) and WinGX (Farrugia, 1999).

This study was financially supported by the Research Centre of Erciyes University. HÜ and MS thank Drs Hasan Seçen and Cavit Kazaz for helpful discussions.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: TY1001). Services for accessing these data are described at the back of the journal.

## References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). J. Chem. Soc. Perkin Trans. 2, pp. S1–19.
- Altomare, A., Burla, M. C., Camalli, M., Cascarano, G., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). J. Appl. Cryst. 32, 115–119.
- Chantegrel, B. & Gelin, S. (1978). J. Heterocycl. Chem. 15, 1215-1219.
- Farrugia, L. J. (1997). J. Appl. Cryst. 30, 565.
- Farrugia, L. J. (1999). J. Appl. Cryst. 32, 837-838.
- Gelin, S. & Gelin, R. (1979). J. Org. Chem. 44, 808-810.

- Gossauer, A. (1983). Tetrahedron, 39, 1933-1941.
- Jimenez, M. D., Ortega, R., Tito, A. & Farina, F. (1988). *Heterocycles*, 27, 173– 183.
- Koz'minykh, V. O., Igidov, N. M., Berezina, E. S., Shavkunova, G. A. & Yakovlev, I. B. (1996). *Pharm. Chem. J. (Engl. Transl.)*, **30**, 458–462.
- Koz'minykh, V. O., Igidov, N. M., Koz'minykh, E. N., Semenova, Z. M. & Andrechikov, Yu. S. (1992). *Pharmazie*, 47, 261–263.
- Koz'minykh, V. O., Igidov, N. M., Zykova, S. S., Kolla, V. E., Shuklina, N. S. & Odegova, T. F. (2002). *Pharm. Chem. J.* **36**, 188–191.
- Nardelli, M. (1995). J. Appl. Cryst. 28, 659.
- Öztürk, S., Saçmacı, M., Üngören, H., Akkurt, M., Fun, H.-K. & Akçamur, Y. (2003). Acta Cryst. E**59**, 0881–0882.
- Parkin, S., Moezzi, B. & Hope, H. (1995). J. Appl. Cryst. 28, 53-56.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Siemens (1996). SMART and SAINT. Versions 4.0. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
- Weigele, M., Blount, J. F., Tengi, J. P., Czajkowski, R. C. & Leimgruber, W. (1972). J. Am. Chem. Soc. 94, 4052–4058.
- Weigele, M., Tengi, J. P., De Bernardo, S., Czajkowski, R. & Leimgruber, W. (1976). J. Org. Chem. 41, 388–393.
- Wiedhopf, R. M., Trumbull, E. R. & Cole, J. R. (1973). J. Pharm. Sci. 62, 1206– 1207.